

Lahore College for Women University Lahore



Lecture:16

Course: Nanotechnology & Nanostructures

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Nanoproduct Forms

Nanoproduct Forms

- Nanocoatings
- Multilayers and Nanofilms
- ❖ Nanoadhesives
- ❖ Nanoporous

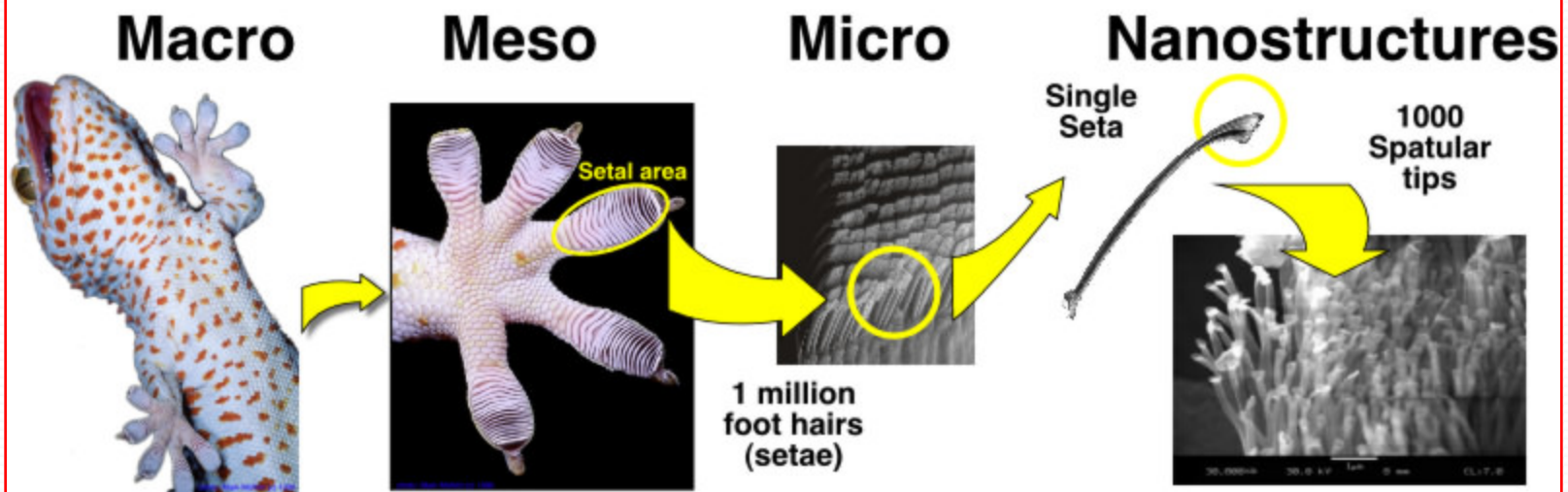
Nanoadhesives

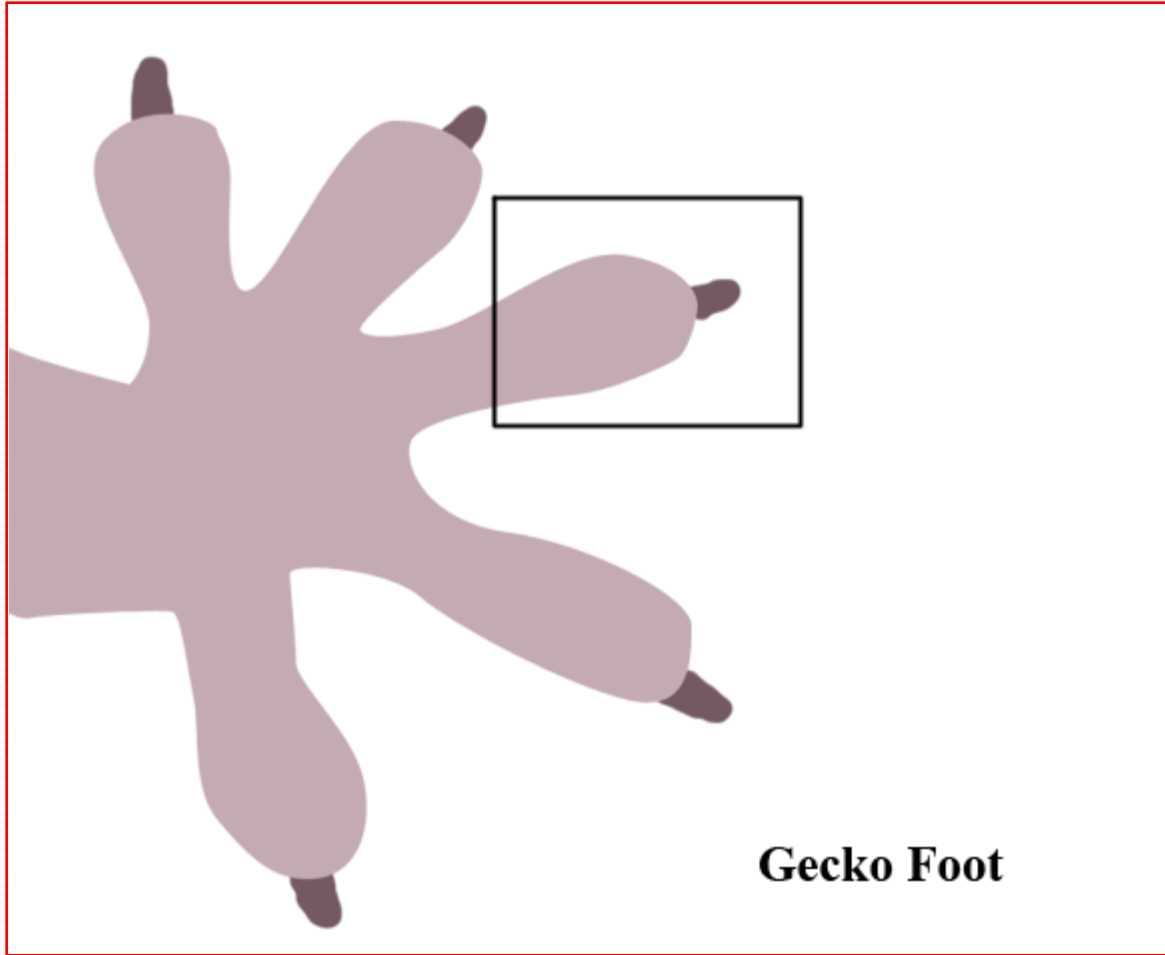
- **Adhesives** are substances that join two surfaces together.
- Historically, many **bulk adhesives** were made from **natural materials**, including plant resins, gums, animal glues and many other substances. **Glues** were made from a wide variety of both organic and inorganic substances (fish, rubber, caseins, and minerals).
- **Adhesives** are used in everything from simple consumer products to automobiles. **Adhesive layers** are relatively **thin**.
- In **Nano-adhesives** those are **bio-inspired, mechanisms** of adhesion are primarily either **mechanical or chemical**.
- **Mechanical means** involve the adhesive substance, providing a **type of interlocking mechanism**, typically by the substance **filling small pores** in each of the surfaces.
- **Chemical means** are more varied and include **direct chemical bonding** and the **development of intermolecular forces**.

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- Corresponding to these several mechanisms, various types of adhesives are
 1. **Reactive adhesives**
 2. **Light-sensitive adhesives**
 3. **Contact adhesives**
 4. **Simple solvent-based adhesives**

- **Nanoadhesive** approaches are those inspired by **nature**, particularly **geckos** and **mussels**.
- **Gecko** has the ability to scamper up walls and even upside down across ceilings.
- This clinging ability, however, **does not come** from any kind of **natural glue**. Rather, it results from the presence of many small hairs that cover their feet. These hairs are much smaller than human hairs and are then divided into even smaller hairs at their ends (**called setae**). At the ends of these multiple tiny split hairs are **spatulae** that are cupped like and around in size. These spatulae radically **increase the contact area between the hairs and a surface**. These spatulae range from the high millions to the low billions. Adhesion results from **intermolecular forces** between the spatulae and the surface, **including Van der Waals' forces** and some **capillary action**.
- They are relatively **weak forces**, but the **high contact area** of the spatula results in an **appreciable** overall **sticking force**.

Gecko adhesive system





Gecko Foot

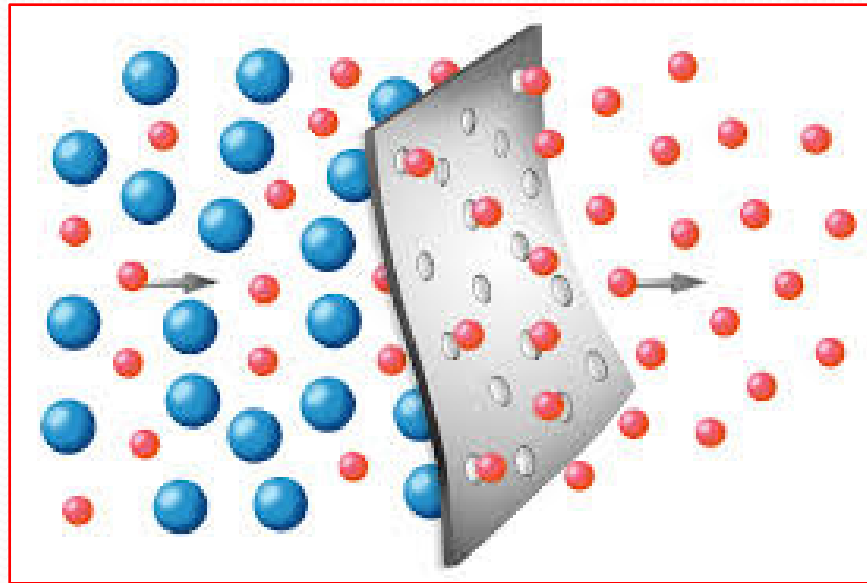
- Several approaches have been explored.
- Most use some type of nanosurface that is patterned after those of a **gecko's foot soles**. But this sticking power decreases in the presence of **high humidity** and often **did not work** on **wet surfaces**.
- Investigators working on “**wet adhesion**” inspired from nature. Here a primary model was the **mussel**, which can be seen in **water environments** sticking to seemingly everything, from rocks to pilings.
- Studies of mussels demonstrated that they secrete specialized proteins (“**DOPA**”) that provide these unique adhesive qualities.
- Now investigators reported that by **coating gecko-mimicked nanopatterns** with a **mussel mimicked polymer** containing the needed **protein**, adhesive qualities could be greatly improved.

Nanoporous

- Nanoporous materials consist of a regular **organic or inorganic framework** supporting a **regular, porous structure**.
- The **size** of the pores is generally **100 nanometers or smaller**.
- Most nanoporous materials can be **classified** as
 1. **Nanoporous Membranes**
 2. **bulk nanoporous materials**

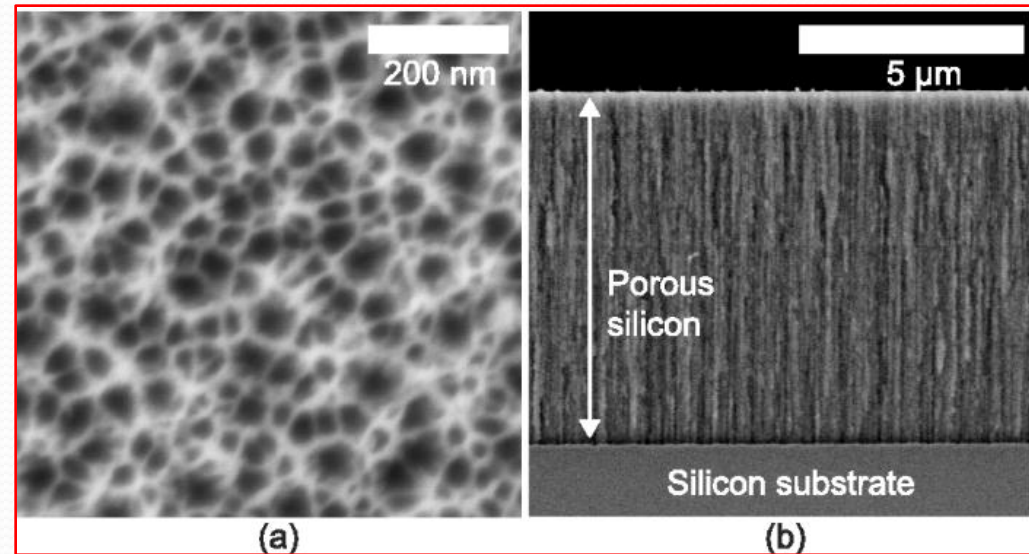
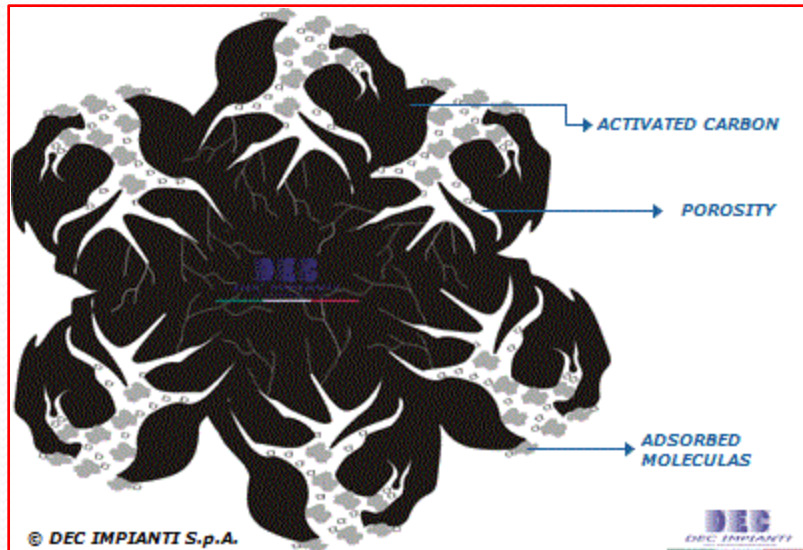
Nanoporous Membranes

- Cell membranes can be thought of as nanoporous membranes.

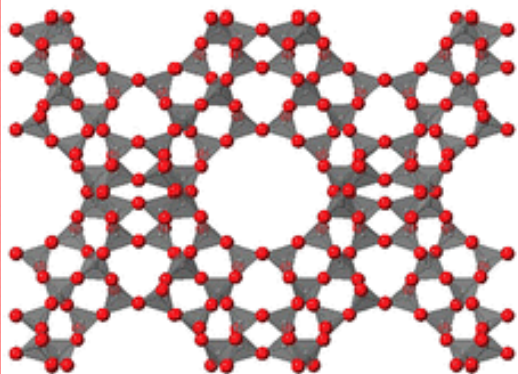


Bulk Nanoporous

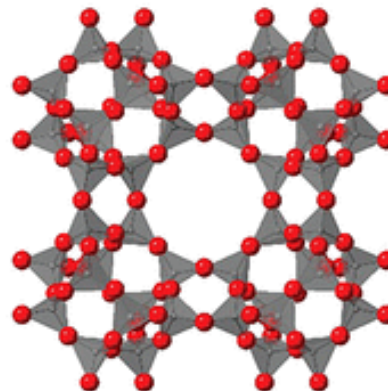
- **Activated carbon, nanoporous silicon and zeolites** (zeolite” means “boiling stone) are examples of bulk nanoporous materials.
- The petroleum industry has been using naturally occurring zeolites as catalysts for decades.



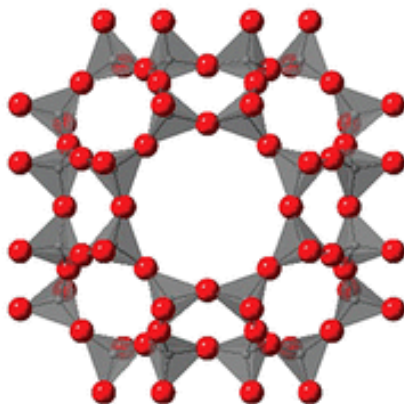
zeolite Na-X (FAU)
 $Fd\bar{3}m$



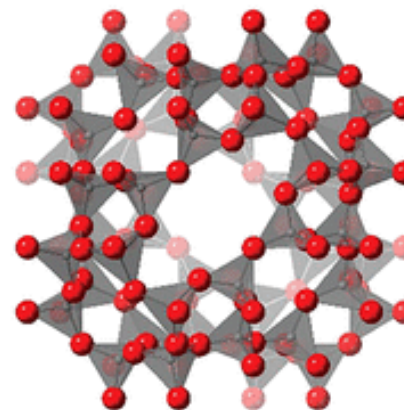
zeolite ZK-5 (KFI)
 $Im\bar{3}m$



zeolite RHO (RHO)
C-form- $Im\bar{3}m$



zeolite RHO (RHO)
A-form- $I4\bar{3}m$



- There are many natural nanoporous materials, but artificial materials can also be manufactured.
- **Processing methods** generally fall into **three** primary categories:
 1. methods that employ **thermal or mechanical means only**,
 2. methods that use some type of **pore-generating agent**,
 3. methods that use some form of **template (molecular imprinting or micellar imprinting)**

- A nanoporous material with consistently sized pores has the property of **letting only certain substances** pass through, while **blocking others**.
- With nanoporous materials, as **pore sizes** (diameters) **decrease**, the **relative pore surface area** radically **increases**.

Subdivisions OF Nanoporous Materials

- Nanoporous materials can be subdivided into 3 categories:
 1. Microporous materials: 0.2–2nm pores
 2. Mesoporous materials: 2–50nm pores
 3. Macroporous materials: 50–1000nm pores

Uses

- Porous materials are used in many fields with different industrial applications:
- They serve as **permeable membranes** for **filtering** or other functions.
- Nanoporous materials are being used as **catalysts**.
- Nanoporous materials are being developed as **breathable films** for **product packaging** that have useful chemical and physical properties.

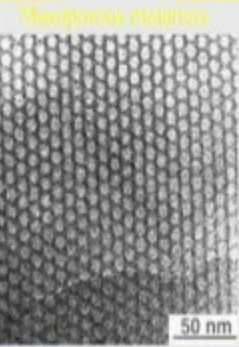
Applications

Environment field

adsorbents, catalysts

Biomedical field

drug delivery, bone regeneration



Chemical catalysis

catalysts, supports

Nanomaterials preparation

nanoparticles, nanoarray

Functional devices

sensors, solar cells



Thank You